

NOTE

Hydrothermal Synthesis and Crystal Structure of 1D Zigzag Chain $[\text{Zn}(\text{Hbsal})_2(4,4'\text{-bipy})]_n$

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A novel coordination polymer of $[\text{Zn}(\text{Hbsal})_2(4,4'\text{-bipy})]_n$ (Hbsal = 5-bromosalicylate, 4,4'-bipy = 4,4'-bipyridine) was synthesized by hydrothermal reaction and characterized by elemental analysis and X-ray single crystal diffraction. The crystal is monoclinic, space group C2/c with $a = 20.561(4)$, $b = 9.207(2)$, $c = 14.769(3)$ Å, $\alpha = 90^\circ$, $\beta = 121.92(3)^\circ$, $\gamma = 90^\circ$, $M_r = 653.58$, $V = 2373.0(8)$ Å³, $D_c = 1.829$ g/cm³, $F(000) = 1288$ and $Z = 4$. The structure analysis shows that the complex exhibits a 1D zigzag chain structure.

Key Words: Hydrothermal, Synthesis, Crystal structure, Zn(II), Coordination polymer.

Salicylic acid and its substituted derivatives continue to attract attention because of its versatile coordination modes and biological applications¹⁻⁵. Many complexes with salicylic acid and N-donor ligands, such as 2,2'-bipyridine, 1,10-phenanthroline and 4,4'-bipyridine, were found to display diverse structure types⁶⁻⁹. Herein, we report the hydrothermal synthesis and crystal structure of a Zn(II) complex, $[\text{Zn}(\text{Hbsal})_2(4,4'\text{-bipy})]_n$.

All chemicals were of AR grade and used as received from commercial sources. Elemental analyses were conducted on a Perkin-Elmer 2400 CHN elemental analyzer.

Synthesis of $[\text{Zn}(\text{Hbsal})_2(4,4'\text{-bipy})]_n$: A mixture of $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ (0.030 g, 0.1 mmol), 4,4'-bipy (0.016 g, 0.1 mmol), 5-bromosalicylic acid (0.043 g, 0.2 mmol) and distilled water (10 mL) was put into a Teflon-lined autoclave (20 mL) and then heated at 433 K for 48 h. Yellow block-like crystals of the complex formed. Yield 30 % (based on Zn). Anal. calcd. (%) for $\text{C}_{24}\text{H}_{16}\text{N}_2\text{O}_6\text{Br}_2\text{Zn}$: C, 44.10; H, 2.47; N, 4.28. Found (%): C, 44.30; H, 2.35; N, 4.35.

X-Ray crystallography: A yellow block-like single crystal with dimension of 0.18 mm × 0.15 mm × 0.14 mm for $[\text{Zn}(\text{Hbsal})_2(4,4'\text{-bipy})]_n$ was used for X-ray diffraction analysis. Data collection was carried out at 293 K on a Rigaku RAXIS-RAPID Weissengberg IP diffractometer with graphite monochromated MoK_α radiation ($\lambda = 0.71073$ Å). A total of 9271 reflections were obtained and 2147 unique ($R_{\text{int}} = 0.0516$) were collected in the range of $3.25 < \theta < 25.25^\circ$ by ω scan

mode, of which 1408 reflections with $I > 2\sigma(I)$ were used in the succeeding refinement. The final $R = 0.0546$, $wR = 0.1303$ ($w = 1/[\sigma^2(F_o^2) + (0.0599P)^2 + 9.4277P]$, where $P = (F_o^2 + 2F_c^2)/3$). The highest and lowest residual peaks in the final difference Fourier map are 1.125 and -0.907 e Å⁻³, respectively. All calculations were performed by the SHELXTL 97 program¹⁰. The selected bond lengths and bond angles are listed in Table-1. CCDC: 786194.

TABLE-1
SELECTED BOND LENGTHS (Å) AND
ANGLES (°) OF $[\text{Zn}(\text{Hbsal})_2(4,4'\text{-bipy})]_n$

Zn1-O1	2.465 (5)	Zn1-N1	2.072(5)
Zn1-O2	2.027 (5)	O2 ^a -Zn1-O1	99.33 (2)
O2-Zn1-O2 ^a	149.2(3)	N1-Zn1-N1 ^a	104.8 (3)
O2-Zn1-N1	94.1(2)	N1-Zn1-O1	150.0(2)
O2 ^a -Zn1-N1	104.7(2)	O1-Zn1-O1 ^a	87.2 (2)
O2-Zn1-O1	56.8 (2)	N1 ^a -Zn1-O1	91.0(2)

Symmetry codes a: $-x + 2, y, -z + 3/2$.

The local coordination around the Zn(II) atom is shown in Fig. 1. The Zn(II) atom in $[\text{Zn}(\text{Hbsal})_2(4,4'\text{-bipy})]_n$ is coordinated by four O atoms from two 5-bromosalicylate ligands and two N atoms from two 4,4'-bipy ligand in a distorted octahedral coordination geometry. All carboxylic anions of 5-bromosalicylate ligands are bidentately coordinated with Zn(II). Through 4,4'-bipy ligands, the 1D zigzag chain along the a axis is generated shown in Fig. 2.

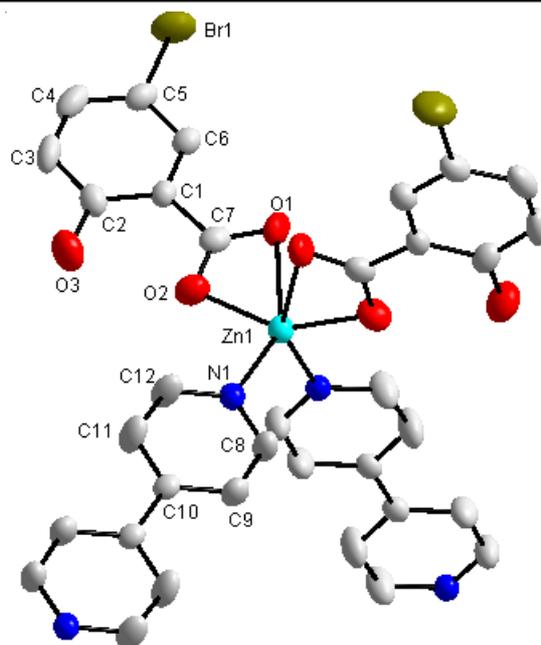


Fig. 1. ORTEP view of coordination environment of $[Zn(Hbsal)_2(4,4'-bipy)]_n$ with 30 % probability ellipsoid

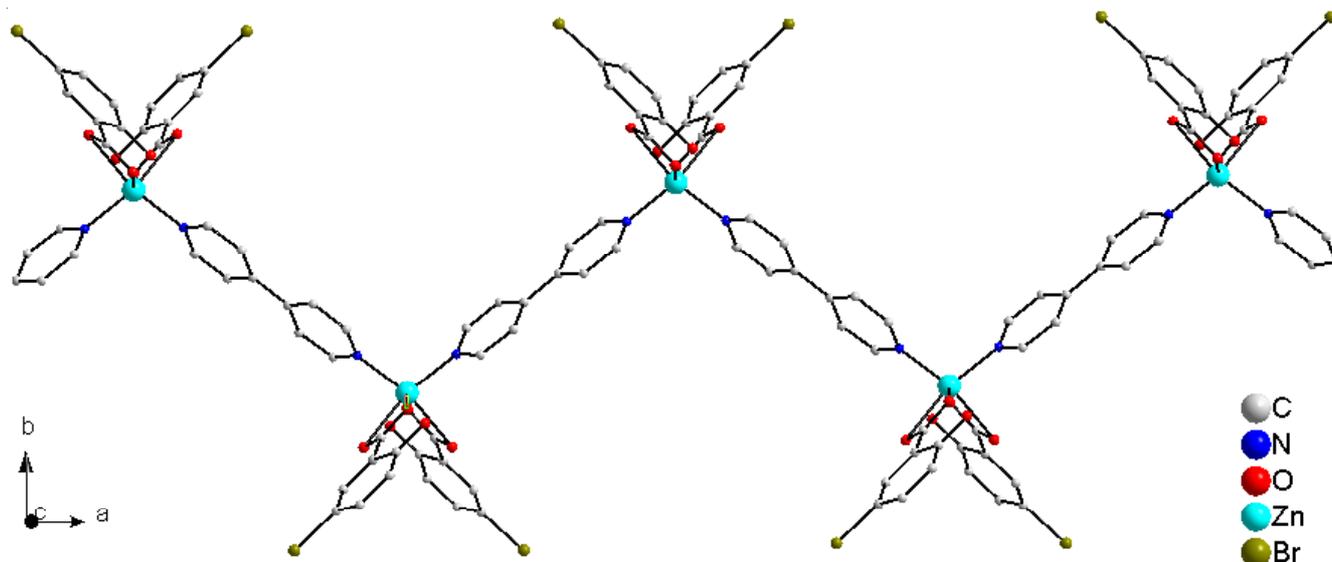


Fig. 2. View of the 1D zigzag chain along the a axis

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